# St. Mary's School, Dwarka <br> Holidays Homework <br> Class: XI <br> Subject: Chemistry <br> Week 1 <br> Worksheet 1 

## Objective:

Ø Revision of concepts
Ø Application of the concepts to real life situations.
Ø Skills to carry out research work and develop scientific aptitude

## Instructions:

*Neatly write all the answers in your science notebook.
*Attempt the questions keeping in mind the weightage of each question.
*Assignment ‘Summer Holiday Homework' will be created on TEAMS. PDF of handwritten work should be uploaded on it.

## Research Project

Observe the packed food item at your home. Find out the kind of packaging materials used. Categorised them into metal based and other materials. Deduce a suitable method to either decompose them at home or reuse in home.

## Assignment 1

Assignment (Determination of pH )

1. Define pH of a solution.
2. Which one of the following solutions has the higher pH ?
(a) $0.1 \mathrm{M} \mathrm{CH}_{3} \mathrm{COOH}$
(b) $0.1 \mathrm{M} \mathrm{HNO}_{3}$
(c) 0.1 M HCl
(d) $0.1 \mathrm{M} \mathrm{H}_{2} \mathrm{SO}_{4}$
3. What is the pH of pure water at $25^{\circ} \mathrm{C}$ ?
4. What is the pH of 0.01 M HCl solution? 2
5. What is the pH of a $1 \times 10^{-5} \mathrm{M} \mathrm{NaOH}$ solution?
6. Which one of the following is true about an acidic solution?
$\left[\mathrm{H}^{+}\right]>10^{-7}$ or $\mathrm{pH}>7$ or $\left[\mathrm{H}^{+}\right]=\left[\mathrm{OH}^{-}\right]=10^{-7} 2$
7. What does the pH of a solution signify? 2
8. The pH of an alkaline solution is 10 . What is its pOH ? 3
9. pH of a solution of $\mathrm{HNO}_{3}$ is 4 . What is the concentration of hydrogen ions? 3
10. What is the relationship between pH and pOH of an aqueous solution? 3
