## St. Mary's School, Dwarka Holiday Homework Class XII Subject: Chemistry Week 4 Worksheet 4

## **Objective:**

- Ø Revision of concepts
- Ø Application of the concepts to real life situations.
- Ø Skills to carry out research work and develop scientific aptitude

## **Instructions:**

- \*Neatly write all the answers in your science notebook.
- \*Attempt the questions keeping in mind the weightage of each question.
- \*Assignment 'Summer Holiday Homework' will be created on TEAMS. PDF of handwritten work should be uploaded on it

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<u>•</u> Q1.	Illustrate the Gabriel phthalimide synthesis.	1
Q2.	Give chemical equations for the conversions of Nitrobenzene to benzoic acid.	1
Q3.	Account for the following:	
	(i) Electrophilic substitution in aromatic amines takes place more readily than benzene.	
	(ii) Nitro compounds have higher boiling points than hydrocarbons having almost same	
	molecular mass.	2
Q4.	(a) Write the product in the following reaction	
	CH <sub>3</sub> CH =CHCH <sub>2</sub> CN (1) DIBAL-H (2)H <sub>2</sub> O $\rightarrow$	
	(b) How can you distinguish between propanal and propanone?	2
Q5.	(a) Why is benzene diazonium chloride not stored and used immediately after its preparation?	
	(b) Diazonium salts of aromatic amines are more stable than those of aliphatic amines.	
	Explain.	2
Q20.	Vedanta is a student of Chemistry. One day, his supervisor has sent him to perform an	
	experiment for the preparation of primary and secondary amines using alkyl halide as a	
	starting material. He started the experiment and completed the reaction properly but he did	
	not get the good yield of primary and secondary amines. He choose the another method and	
	found the desired result.	
	Answer the following questions.	
	(i) Which type of reaction was going to perform by Vedanta?	
	(ii) Write all the possible products of the above reaction.	
	(iii) Can you choose more precise method for the preparation of primary amine as a product	
	mainly?	3
Q21.	Write short note on the following:	
	(i) Carbylamine reaction	
	(ii) Hoffmann's bromide reaction	
	(iii) Coupling reaction	3

Q24. (a) Write the chemical equation for the reaction involved in cannizzaro reaction. (b) Draw the structure of semicarbazone of ethanal (c) Why pka of F-CH<sub>2</sub>COOH is lower than that of Cl-CH<sub>2</sub>COOH 3 Q25. Given reasons: (i) Acetylation of aniline reduces its activatino effect. (ii) CH<sub>3</sub>NH<sub>2</sub> is more basic than C<sub>6</sub>H<sub>5</sub>NH<sub>2</sub> (iii) Although –NH<sub>2</sub> group is o/p directing, yet aniline on nitration gives a significant amount of m-nitroaniline. 3 Q19. Arrange the following: (i) In decreasing order of pKb values:  $C_2H_5NH_2$ ,  $C_6H_5NHCH_3$ ,  $(C_2H_5)_2NH$  and  $C_6H_5NH_2$ (ii) In increasing order of basic strength: (a) Aniline, p-nitroaniline and p-toluidine (b) C<sub>6</sub>H<sub>5</sub>NH<sub>2</sub>, C<sub>6</sub>H<sub>5</sub>NHCH<sub>3</sub>, C<sub>6</sub>H<sub>5</sub>CH<sub>2</sub>NH<sub>2</sub> (iii) In decreasing order of basic strength: C<sub>6</sub>H<sub>5</sub>NH<sub>2</sub>, C<sub>6</sub>H<sub>5</sub>N(CH<sub>3</sub>)<sub>2</sub>, (C<sub>2</sub>H<sub>5</sub>)<sub>2</sub>NH, CH<sub>3</sub>NH<sub>2</sub> (iv) Decreasing order of basic strength in gas phase: C<sub>2</sub>H<sub>5</sub>NH<sub>2</sub>, (C<sub>2</sub>H<sub>5</sub>)<sub>2</sub>NH, (C<sub>2</sub>H<sub>5</sub>)<sub>3</sub>N and NH<sub>3</sub> (v) Increasing order of boiling point: C<sub>2</sub>H<sub>5</sub>OH, (CH<sub>3</sub>)<sub>2</sub>NH, C<sub>2</sub>H<sub>5</sub>NH<sub>2</sub> 5

